Microwave Spectrum of Ethylene Oxide in Excited Vibrational States

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In the rotational spectrum of ethylene oxide, the five lowest-lying vibrational states are assigned and the vibration-rotation constants and the centrifugal distortion constants are derived. The vibrational frequencies are estimated from relative intensity measurements at $v_a=803\pm18~\rm cm^{-1}$ and $v_b=814\pm23~\rm cm^{-1}$ for the antisymmetric species (with respect to the C_2 operation) and at $v_c=872\pm34~\rm cm^{-1}$, $v_d=1020\pm33~\rm cm^{-1}$, and $v_e=1109\pm42~\rm cm^{-1}$ for the symmetric species. The intensity measurements of the remaining spectral lines showed that these lines belong to vibrational states at frequencies above about $1000~\rm cm^{-1}$. The v_a , v_b , and v_c states are tentatively assigned to v_{15} , v_{12} , and v_5 , respectively, by taking into account the Coriolis interactions.

With a considerable number of infrared and Raman investigations carried out so far, $^{1-13}$) the assignments of certain fundamental frequencies of ethylene oxide are still subject to controversy. Namely, as shown by some examples in Table 1, different authors have proposed considerably different frequencies for v_7 (A_2 , CH_2 -twisting), v_8 (A_2 , CH_2 -rocking), and v_{12} (B_1 , ring deformation), for which no direct spectral evidence has been obtained. Under these circumstances, further infrared and Raman studies would seem to be ineffective in improving the situation significantly, and rather somewhat different approach is required.

Now, it is known that the intensity measurements of microwave absorption in vibrationally-excited states¹⁴⁾ and the analysis of the vibration-rotation interaction^{15–18)} provide useful information about the energy and symmetry of these vibrational states. In fact, these methods may certainly be advantageous

Table 1. Vibrational assignment of ethylene oxide (in cm⁻¹)

Syn	nmetry	Type of vibration	Lord and Nolin ^{a)}	Potts ^{b)}	Hirokawa et al.c)	Cant et al.d)
$\overline{A_1}$	$v_1(CH)$	stretch)f)	3005	(3006) e	3005	3018
	$v_2(CH$	deform)	1490	1497.5	1494	1498
	v_3 (ring	breath)f)	1266	1270.5	1270	1270
	$v_4(CH)$	wag)f)	1120	$(1130)^{e}$	1120	1148
	$v_5(\text{ring})$	deform)	877	877	877	877
A_2	$\nu_6(\mathrm{CH}$	stretch)	3063	$(3065)^{e}$	3065	3073g)
_	$v_7(CH$	2 twist)	$(1345)^{e}$	(1300)	1030	1047g)
	$\nu_8({ m CH}$	2 rock)	807	$(860)^{e}$	807	825g)
B_1	$\nu_{9}(\mathrm{CH}$	stretch)f)	3019	3006	3019	3006
	$v_{10}({ m CF}$	I ₂ deform)	1470	1471.5	5 1470	1472
	$v_{11}(CF)$	I ₂ wag)	1153	1151	1153	1151
		g deform)	892	$(890)^{6}$	840	840
B_2	$v_{13}(\mathrm{CF}$	I stretch)	3079	3065	3065	3065
	$v_{14}({ m CF}$	I ₂ twist)	1143	1142	1143	1142
	$v_{15}(CF$	$\mathbf{I_2} \operatorname{rock}$	821	821.5	5 822	821

a) Ref. 5. b) Ref. 6. c) Ref. 12. d) Ref. 13. e) Values in parenthesis were estimated in the original papers. f) v_1 , v_3 , v_4 , and v_9 respectively, are assigned to the bands at 3018, 1269, 1123, and 3008 cm⁻¹ of the Raman spectra of gaseous C_2H_4O obtained by the present authors. g) The values from IR spectra of solids.

when the vibrational transition is extremely weak or forbidden as in those described above. Thus, the present work was carried to out examine the frequencies and the assignments of the lower fundamental frequencies of ethylene oxide using microwave spectroscopy.

Experimental

Ethylene oxide was purchased from the Matheson Co. and was used without further purification. The rotational spectrum was observed with a conventional spectrometer at 300 K mainly in the frequency region from 9 to 42 GHz. Previous results $^{19-21)}$ were referred to in eliminating the spectral lines of isotopic species. A double resonance experiment was performed on the $2_{11} \leftarrow 2_{02}$ and $2_{20} \leftarrow 2_{11}$ transitions in order to confirm the assignment of the Q-branch transitions. Spectral line intensities were measured from a chart recorded using a Hewlett-Packard MRR8460A-607 R-band spectrometer. $^{24)}$ The accuracy of the measurements was checked by measuring the natural abundance ratio of $^{12}\mathrm{C}^{13}\mathrm{CH_4O}$ to $^{12}\mathrm{C}_2\mathrm{H_4O}$.

Results and Discussion

The microwave spectra of ethylene oxide and its isotopic species have been investigated in the ground state by a number of authors, 19-23) resulting in the determination of the r_0 , r_s , and r_m structures, 19-21) the dipole moment, 19) the centrifugal distortion constants, 21-23) and the 17O quadrupole coupling constants.23) On the basis of these results, the rotational spectrum of the vibrationally-excited states was identified and frequencies were assigned to the transitions in five low-lying vibrational states, v_a , v_b , v_c , v_d , and v_e in increasing order of frequency, as listed in Table 2. Since the molecular dipole moment lies along the b-axis, a rotational transition $J_{K-1K_1} \leftrightarrow$ $J_{K_{-1}'K_{1}'}$ is allowed only for such $K_{-1}K_{1} \leftrightarrow K_{-1'}K_{1'}$ combinations as ee↔oo and eo↔oe, where e and o signify even and odd. The calculated transition frequencies for the three lower vibrational states, v_a , v_b , and v_c , were obtained using the parameters given in Table 3, which were determined by a leastsquares fit including the centrifugal distortion effect to the first order. 25) For $\nu_{\rm d}$ and $\nu_{\rm e}$ states no R-branch transitions were identified, so that their calculated transition frequencies were derived from the observed values of $l_{10} \leftarrow l_{01}$ and $2_{20} \leftarrow 2_{11}$ via Q-branch plots without taking account of the centrifugal distortion.

In Table 4, the intensities of the excited state lines

Table 2. Rotational transitions of ethylene oxide in excited vibrational states (in MHz)

Turnition	v _a state		$v_{\rm b}$ state		$v_{\rm e}$ state		$v_{\rm d}$ state		ν _e state				
Transition	Obsd	Obsd – Calcda	Obsd	Obsd – Calcda)	Obsd	Obsd-	– Calcda)	Obsd	Obsd	-Calcde)	Obsd	Obsd	–Calcd ^{c)}
111-000	39464.1	1 - 0.00	38849.0	4 - 0.00	39939.	20	0.00						
$2_{02} - 1_{11}$	63135.78	-0.24	62460.2	6 0.25	64345.0	03	0.18						
$2_{12} - 1_{01}$	67484.20	0.47	66105.4	-0.10	68771.	68 –	0.36						
$2_{21} - 1_{10}$	90368.4	4 - 0.22	89292.5	1 - 0.15	90981.	28	0.18						
$1_{10} - 1_{01}$	11442.8	2 0.01	11594.6	2 - 0.11	11104.	31 –	0.13	11206.	23	0 00	10746	.71	0.00
$2_{11} - 2_{02}^{b)}$	24875.3	3 0.04	26462.8	3 - 0.02	23818.	16	0.13	24793.	.06	-0.89	24795	.98	8.83
$2_{21} - 2_{12}$	34327.7	4 0.00	34781.8	5 - 0.03	33313.	46 –	0.04	33618.	.01	-0.68	32246	.05	5.92
$2_{20} - 2_{11}^{\text{b}}$	15790.6	8 - 0.48	15238.7	7 0.21	15531.	11 -	0.05	15194.	.78	-0.02	13978	.02	0.00
$3_{21} - 3_{12}$			24869.1	2 - 0.12	22693.	94 -	0.14	23427.	.70	-1.70	-	_	
$3_{31} - 3_{22}$	40008.9	9 - 0.03	39634.3	0 - 0.14	39061.	58 –	0.07	38864.	.19	-2.34	_	-	
$3_{30} - 3_{21}$	23561.0	9 - 0.01	21692.2	9 0.26	23454.	24 -	0.07	22304.	.39	-0.67	19695	.59	11.01
$4_{22} - 4_{13}$	41406.5	8 - 0.05		-	39439.	80	0.01	41517.	.05	-5.06	-		
$4_{31} - 4_{22}$	25002.3	5 0.31	25211.1	5 0.31	24361.	27	0.10	24395.	.55	-2.96	23411	.99	9.81
$4_{40} - 4_{31}$	34884.0	0 - 0.06	31300.3	9 - 0.24	34919.	31	0.20	32754	.56	-3.69	-	-	
$5_{32} - 5_{23}$	37613.4	4 0.15	40960.6	0.16					-	-	38658	.26	31.06
$5_{41} - 5_{32}$	30155.7	3 0.37	28511.5	-0.07	29917.	02	0.06	28750.	.93	-5.80	26126	.32	24.97
$6_{42} - 6_{33}$			38061.8	-0.11	34407.	62 –	0.15	35560	.69	-9.55	35836	.07	25.12
$6_{51} - 6_{42}$	39574.3	0.03	35406.3	0.29	39786.	69 –	0.18	37052	.08 –	12.10	32037	.08	78.42
$7_{52} - 7_{43}$	37680.9	7 - 0.25	37669.2	6 0.05	36967.	51	0.16	36549	.65 –	14.51	35080	.86	33.86
$8_{62} - 8_{53}$			40927.9	08 -0.15				41765.	.74 –	-29.90	37492	2.10	91.00

a) Errors in the observed values are generally within ± 0.1 MHz. b) The transitions were confirmed by a double-resonance experiment. c) The calculated frequencies of the rotational transitions for the $v_{\rm d}$ and $v_{\rm e}$ states were derived from the $1_{10}-1_{01}$ and $2_{20}-2_{11}$ transitions using Q-branch plots without taking account of the centrifugal distortion.

Table 3. Rotational and centrifugal distortion constants in excited vibrational states (in MHz)a1

	Ground stateb)	$ u_{\mathbf{a}} $	$ u_{\mathbf{b}} $	$ u_{\mathbf{c}}$
$A_{ m s}$	25483.92	25453.45	25222.09	25521.92
$B_{\mathbf{s}}$	22120.82	22003.60	22250.62	22041.25
$C_{\mathbf{s}}$	14097.82	14010.75	13626.91	14417.47
α_s^a		30.47	261.83	-38.00
$\alpha_s^{\ b}$		117.22	-109.80	79.57
$\alpha_s{}^c$		87.07	470.91	-319.65
$ au_{aaaa}'$	-0.2832 ± 0.0084	-0.29 ± 0.27	-0.30 ± 0.20	-0.36 ± 0.17
$ au_{ m bbbb}{}'$	-0.1250 ± 0.0085	-0.01 ± 0.28	-0.29 ± 0.21	-0.21 ± 0.18
$ au_{ m cccc}{}'$	-0.0327 ± 0.0086	-0.06 ± 0.30	0.46 ± 0.23	-0.43 ± 0.19
t ₁ c)	-0.2351 ± 0.0110	-0.59 ± 0.37	0.11 ± 0.26	-0.35 ± 0.23
t ₂ °)	-0.1021 ± 0.0144	-0.91 ± 0.49	0.65 ± 0.37	-0.29 ± 0.31
	Ground stateb)	$v_{ m d}$		$ u_{\mathbf{e}} $
$2B_{\rm s}-A_{\rm s}-C_{\rm s}$	4659.90	4799.59		5404.57
$A_{\rm s}-C_{\rm s}$	11386.10	11206.23		10749.00

a) Error limits for the rotational constants and the vibration-rotation constants (α_s^x) for v_a , v_b , and v_c were estimated from 2.5 times the standard deviations to be less than ± 0.5 MHz. For the v_d and v_e states the values of $2B_s - A_s - C_s$ and $A_s - C_s$ were derived using Q-branch plots of the observed values of the $l_{10} \leftarrow l_{01}$ and $l_{20} \leftarrow l_{11}$ transitions, and the reliability intervals were estimated to be $l_{10} \leftarrow l_{11} = l_{11} =$

are given by the ratios to those of the corresponding ground state lines. The table is limited within the frequency range covered by the MRR8460A-607 R-band spectrometer, and leaves quite a few lines unlisted because of the uncertainty due to overlapping.

It can be noticed, however, that in the $v_{\rm a}$ and $v_{\rm b}$ states the relative intensity Γ of eo \leftrightarrow oe type transitions is about three times that of ee \leftrightarrow oo type values, while the Γ is approximately equal for both types of transitions in the $v_{\rm e},\ v_{\rm d}$, and $v_{\rm e}$ states.

TABLE 4.	Values of $\exp(-hv/kT)\cdot g^{\mathrm{v}}(J_{K_{-1}K_1})/g^0(J_{K_{-1}K_1}),$ a) the Boltzmann factor,
	AND ESTIMATED VIBRATIONAL FREQUENCIES

	v_{a}	$v_{ m b}$	$ u_{\mathbf{c}}$	$ u_{ m d}$	$v_{ m e}$	$^{12}\mathrm{C}^{13}\mathrm{CH_4O}$
$e^{-h u/kT}\cdot g^{ abla}(J_{\mathrm{ee}})/g^{0}(J_{\mathrm{ee}})$	$I_{\rm ee}$) and/or $e^{-h\nu/kT}$.	$g^{\mathrm{v}}(J_{\mathrm{oo}})/g^{\mathrm{o}}(J_{\mathrm{oo}})$	The state of the s			
111-000	_	0.0120	0.0141			0.0164
$3_{31} - 3_{22}$		_	0.017_{0}	0.007_{9}		
$4_{41} - 4_{31}$	0.012_{6}		0.014_{3}	0.007_{6}		
$6_{42} - 6_{33}$		0.011_{3}	0.015_{0}	0.007_{5}	0.004_{4}	0.016,
$6_{51} - 6_{42}$	$(0.014_4)^{\mathrm{b}}$	0.012,	0.015_{4}	0.007_{6}	0.005_{0}	
Average	0.012_{6}	0.012_{0}	0.015_2	0.007,	0.004_{7}	0.016_6
$e^{-harphi/kT}\!\cdot\! g^{\scriptscriptstyle abla}(J_{ m eo})/g^0(J_{ m eo})$	$J_{ m eo}$) and/or ${ m e}^{-h u/kT}$.	$g^{ m v}(J_{ m oe})/g^{ m 0}(J_{ m oe})$				
$2_{21} - 2_{12}$	0.0351	0.0348	0.0153	0.0066		0.0285
$5_{41} - 5_{32}$	0.034_{8}	0.034_2	0.015_{9}			0.027_{7}
$5_{32} - 5_{23}$	0.037_{5}		_		0.005_{2}	
$7_{52} - 7_{43}$	0.034,	0.033_{6}	0.015_1	0.007_{6}	0.005_{0}	0.029_3
Average	0.035_{5}	0.034_2	0.015_{3}	0.007_{1}	0.005_{1}	0.028_{5}
$e^{-h \nu/kT}$	0.0213	0.0202	0.0153	0.0075	0.0049	0.0211
$2.5\sigma^{c)}$	0.0018	0.002_{1}	0.002_{3}	0.001	0.000_{9}	$0.001_{\scriptscriptstyle 5}$
$v(\mathrm{cm}^{-1})$	803	814	872	1020	1109	
	±18c)	±23c)	±34°)	±33c)	±42°)	

a) $g^0(J_{K-1K_1})$ and $g^v(J_{K-1K_1})$ denote the spin weights of the J_{K-1K_1} rotational level for the ground and excited vibrational states, respectively. b) Since Stark components were not fully resolved, the value was not used in the subsequent analysis. c) 2.5 times the standard deviations introduced by $\exp(-hv/kT) \cdot g^v/g^0$ are indicated.

Now, this fact can be understood as arising from the degeneracy of the hydrogen nuclear spin states. Namely, the symmetry requirement for the composite wave functions formed by the vibrational, rotational, and nuclear spin parts leads to the following statement for the present $C_{2\nu}$ case:²⁶⁾ in the vibrational states of the A₁ and A₂ species, ee and oo type rotationa! levels are associated with a 10-fold nuclear spin weight, while eo and oe levels are associated with a 6-fold weight, with the situation in the B₁ and B₂ vibrational states, however, being inverted, i.e., ee and oo are 6-fold while eo and oe are 10-fold. Then, since the ground state is necessarily an A_1 species, the Γ of the A₁ and A₂ states should be equal to the vibrational Boltzmann factor irrespective of the type of rotational transition. In B_1 and B_2 states, however, the Γ of ee↔oo type and eo↔oe type transitions prove to be equal to 6/10 and 10/6 times the Boltzmann factor, respectively. Thus, it is evident from Table 4 that v_a and v_b are of the B-type species and the rest are of the A-type. The energy of each vibrational state is calculated from the Boltzmann factor obtained as is given in Table 4. Slight changes in the dipole moment, rotational energies, and line widths in different vibrational states were ignored in the above discussion, as they are not expected to influence the conclusion significantly.

Referring to Table 3, it is noted that the rotational constant C and the centrifugal distortion constant $\tau'_{\rm ccc}$ of the $\nu_{\rm b}$ and $\nu_{\rm c}$ states show unusually large deviations from those of the ground state in opposite directions. This observation indicates the presence of a c-type Coriolis interaction between the $\nu_{\rm b}$ and $\nu_{\rm c}$ states. It can be shown that the Coriolis terms in $\alpha^{\rm c}$ and $\tau'_{\rm ccc}$

are approximately given by $\mp 4C^2 (\zeta_{\rm b,c}{}^{\rm (c)})^2/(E_{\rm b}-E_{\rm c})$ and $\mp 64C^4 (\zeta_{\rm b,c}^{(c)})^4/(E_{\rm b}-E_{\rm c})^3$, where the upper and lower signs refer to the v_b and v_c states at the vibrational energies $E_{\rm b}$ and $E_{\rm e}$ respectively. ^{16,18,27)} Taking the conclusion of the preceding paragraph into consideration, the Coriolis coupling constant $\zeta_{b,c}^{(c)}$ proves to vanish except when the v_b and v_c states belong to either B_1 and A_1 or B_2 and A_2 species. Accordingly, after comparing the v_b and v_e energies in Table 4 with those in Table 1, their possible assignments are reduced to the following two cases: (i) v_b to v_{12} (B₁, ring deform) and ν_e to ν_5 (A₁, ring deform), or (ii) $v_{\rm b}$ to v_{15} (B₂, CH₂ rock) and $v_{\rm c}$ to v_{8} (A₂, CH₂ rock). A tentative calculation using the force field fitted to the assignment of Lord and Nolin gives $\zeta_{12,5}^{(c)} = 0.9965$ and $\zeta_{15,8}^{(c)} = 0.3689.^{28}$) Then, the Coriolis parts of α^c and τ'_{ccc} are calculated to be ± 454 MHz and ±0.473 MHz in case (i), and ±62 MHz and ±0.009 MHz in case (ii), respectively. Comparing these results with Table 3, the assignment of case (i) appears to be more probable, though not definite because of the uncertainty in $E_{\rm b}-E_{\rm c}$ value used above. Assuming case (i) above to be correct, v_a , v_d , and v_e may be reasonably assigned to ν_{15} (B₂, CH₂ rock), v_7 (A₂, CH₂ twist) and v_4 (A₁, CH₂ wag), respectively, in view of their frequencies and symmetries (see Table 1). Then the situation arises that the present microwave results lack an assignment for the vibrational state v_8 (A₂, CH₂ rock). As it is very unlikely that the present study should have missed a vibrational state as low as about 800 cm⁻¹,²⁹⁾ this means that the frequency of ν_8 should be higher than those proposed by other authors, as listed in Table 1. In fact, we observed several spectral lines in addition to those

listed in Table 4 in the same frequency region, but their intensities gave little indication of the sixth vibrational state below 1000 cm⁻¹.

This conclusion is in fair agreement with the assignments of Hirokawa et al. 12) and the very recent one by Cant et al.¹³⁾ except for the location of v_8 (A₂). The frequency of the last mode has, so far, always been based on solid spectra obtained at low temperature,7,10,13) for which the infrared prohibition of A2 modes may perhaps be lifted. In solid spectra, however, the appearance of crystal field splittings has always to be taken into account in dealing with such closely lying lines as 825 and 818 cm⁻¹, which have been assigned to ν_8 and ν_{12} by Cant et al. The present result suggests that they are not separate fundamental frequencies but are the split components of v_{12} , contrary to the conclusions in Ref. 13. Incidentally, Cant et al. assigned the corresponding C₂D₄O frequencies at 587 and 779 cm⁻¹, but the former was also very close to the 595 cm⁻¹ line which they assigned to v_{15} . In view of their observation on a dilute $C_2D_4O/$ C₂H₄O mixture, which gave only a single evident line near 590 cm⁻¹, it seems that the lines at 587 and 595 cm⁻¹ are more likely to be the crystal field doublet

Finally, the possibility for case (ii) is mentioned, though it was previously discarded on the basis of Coriolis interaction behavior. Assuming the choice of case (ii), we can conclude that the v_5 (A₁) state is missing in the present result, instead of v_8 . Such a case is very improbable, however, because the v_5 frequency given in Table 1 has been established quite convincingly.

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References

- 1) L. G. Bonner, J. Chem. Phys., 5, 704 (1937).
- 2) K. W. F. Kohlrausch and A. W. Reitz, Proc. Indian Acad. Sci., 8A, 255 (1938).
 - 3) J. W. Linnett, J. Chem. Phys., 6, 692 (1938).
- 4) H. W. Thompson and W. T. Cave, Trans. Faraday Soc., 47, 946 (1951).
 - 5) R. C. Lord and B. Nolin, J. Chem. Phys., 24, 656 (1956).

- 6) W. J. Potts, Spectrochim. Acta, 21, 511 (1965).
- J. LeBrumant and D. Maillard, C. R. Acad. Sci. Paris, **264**, 1107 (1967).
 - 8) J. LeBrumant, ibid., 268, 486 (1969).
 - 9) J. LeBrumant, ibid., 270, 801 (1970).
- 10) V. T. Aleksanyan, E. R. Razumova, A. P. Kurbakova, and S. M. Shostakovskii, Opt. Spectrosc., 31, 369 (1971).
- 11) J. E. Bertie and D. A. Othen, Can. J. Chem., 51, 1155 (1973).
- 12) T. Hirokawa, M. Hayashi, and H. Murata, J. Sci. Hiroshima Univ., 37A, 283 (1973).
- 13) N. W. Cant and W. J. Armstead, Spectrochim. Acta, 31A, 839 (1975).
 - A. P. Cox and A. S. Esbitt, ibid., 38, 1636 (1963).
 - E. Hirota, J. Mol. Spectrosc., 7, 242 (1961). 15)
 - T. Tanaka and Y. Morino, ibid., 32, 436 (1969). 16)
 - V. W. Laurie and D. T. Pence, ibid., 10, 155 (1963). 17)
- E. Hirota and M. Sahara, ibid., 56, 21 (1975).
- G. L. Cunningham, A. W. Boyd, R. J. Myers, W. D. Gwinn, and W. I. LeVan, J. Chem. Phys., 19, 676 (1951). 20) T. E. Turner and J. A. Howe, ibid., 24, 924 (1956).
- C. Hirose, This Bulletin, 47, 1311 (1974).
- 22) C. Hirose, Astrophys. J., 189, L145 (1974).
- 23) R. A. Creswell and R. H. Schwendeman, Chem. Phys. Lett., 27, 521 (1974).
- 24) The present study was initiated by one of the authors (C. H.) during his stay in the laboratory of Professor Robert F. Curl, Jr. at the Department of Chemistry of Rice University, Houston, Texas. The intensities were measured as the line intensities of the spectral lines which were recorded under the same conditions of the sample and spectrometer except that the strength of the Stark field to resolve the Stark components varied from line to line. The preliminary spectrum recorded there was of great help in the subsequent study.
- 25) J. K. G. Watson, J. Chem. Phys., 46, 1935 (1967).
- 26) C. H. Townes and A. L. Schawlow, "Microwave Spectroscopy," McGraw-Hill, New York (1955), p. 101.
- 27) H. H. Nielsen, "Handbuch der Physik," by S. Flügge, Springer, Berlin (1959), p. 173.
- 28) K. Venkateswarlu and P. A. Joseph, Indian J. Pure Appl. Phys., 10, 754 (1972).
- 29) There are two possible causes of missing the rotational lines of a vibrational state as low as about 800 cm⁻¹: (1) overlapping of the lines with those of the ground state, or (2) extremely strong Coriolis interaction which shifts all the transitions outside of the frequency region of the present observation. For the former case, all the three rotational constants of the vibrational state have to be equal to those of the ground state to within a few megahertz, thus, case (1) is unlikely though not impossible. Although, for the latter case, such a strong Coriolis interaction requires a fifth vibrational state located very close to the fourth state, none of the v_a , v_b , and v_c states is perturbed as strongly and no rotational spectra that can be assigned to the fourth and fifth vibrational states were observed in this region. Furthermore, infrared and Raman spectra do not indicate the existence of the fifth state around 800 cm⁻¹. Therefore, this case can be ruled out.